Seat No: _____

Enrollment No:

PARUL UNIVERSITY FACULTY OF PHARMACY

B.Pharm. Summer 2017-18 Examination

Semester: 2 Date: 06/06/2018

Subject Code: 08101151 Time: 10.00 am to 1.00 pm

Subject Name: Pharmaceutical Chemistry-II (Physical Chemistry)

Total Marks: 75

Instructions:

- 1. Figures to the right indicate full marks.
- 2. Make suitable assumptions wherever necessary.

Q.1 Essay type Questions. (Any 2 out of 3) (10 marks each)

(20)

- 1. What are different types of catalysis? Write a note on characteristic of homogenous and heterogeneous catalyst.
- 2. What are Colligative properties? Explain any two in detail.
- 3. What is radioactivity? Explain different types of radiations emitted by radioactive substances. Write a note on Nuclear reaction.

Q.2 Short Essay type Questions. (Any 7 out of 9) (5 marks each)

(35)

- 1. Discuss Debye-Huckel theory in detail.
- 2. Explain principle of Oswald's Viscometer.
- 3. Write a note on carnot cycle.
- 4. Write a detail note on first law of thermodynamics.
- 5. State and explain Roult's law of dilute solution.
- 6. Write a note on Parachlor.
- 7. Draw Jablonski diagram. State and explain Lambert- Beer law of Photochemistry.
- 8. Write a note on Giger Muller counter.
- 9. Enlist various methods for determination of order of kinetics. Discuss any two methods.

Q.3 Answer in short. (2 marks each)

(20)

- 1. Enumerate methods for determination of surface tension.
- 2. Differentiate between Ideal and real solution.
- 3. Define viscosity. Write units used to express viscosity.
- 4. What is phase rule?
- 5. Distinguish between Reversible and irreversible process
- 6. Explain term autocatalytic reaction with example.
- 7. State zeroth law of thermodynamics.
- 8. State and explain Henry's law.
- 9. Prove that half-life of first order reaction is independent of the initial concentration.
- 10. Enlist the Application of radioactivity.