PARUL UNIVERSITY

PARUL INSTITUTE OF APPLIED SCIENCES

MID SEMESTER INTERNAL EXAMINATION, OCTOBER 2017

B. Sc. Semester I

Subject: Biotechnology/Microbiology

Paper Code: 11103101 Date: 10 /10 /2017 Maximum Marks: 40 Instructions: Title of the paper: Biochemistry-1 Time: 10:00-11:30AM

11111e: 10:00-11:50AM

- 1. All questions are compulsory and options are given in first and second question only.
- 2. Numbers to the right of question indicate the marks of respective question.

Q. 1	Attempt any one question of the following.						
	(i) What are the general properties of biomolecules? Discuss in detail.						
	(ii)What are polysaccharides? Discuss any polysaccharide in detail.						
Q. 2	Attempt any three questions of the following.						
	(i) Write the reactions of Kiliani–Fischer synthesis starting from D-						
	ribose						
	(ii) What are bioelements?						
	(iii) What is mutarotation? Explain with the help of an example.						
	(iv) Describe the action of hydrazine on sugars.						
	(v) Explain optical isomerism in Glucose.						
Q. 3	Do as directed. Attempt all five questions.						
	(i) Draw the stuctures of α -D-glucopyranose and β - D-glucopyranose						
	(ii) What is the storage form of carbohydrate in animals and where it is						
	stored?						
	(iii) Name any two biotechnologically important carbohydrates.						
	(iv) List name of any two glycosaminoglycans.						
	(v) Give one example of epimers.						
Q. 4	Write correct option in your answer sheet for following 15 multiple						
	choice questions.						
MCQ 1	Ketoses are reducing sugars because in their non-cyclic form they contain						
	(A)	Aldehyde group	(B)	Ketone group			
	(C)	Hydroxyl group	(D)	Ester group			
MCQ 2	What is the relationship between mannose and galactose?						
	(A)	They are stereoisomers	(B)	They are epimers			
	(C)	They are aldoses	(D)	All of the above			
MCQ 3	Which of the following results when the galactose is reduced						
	(A)	Galacturonic acid	(B)	Galactitol			
	(C)	Galactonic acid	(D)	None			
MCQ 4	What is a storage form of carbohydrate in plants?						
	(A)	Glycogen	(B)	Starch			
	(C)	Cellulose	(D)	Chitin			
MCQ 5	What best describes amylose?						
	(A)	Unbranched linear polymer	(B)	Unbranched linear poly	ner with		
		with alpha 1-4 glycosidic		beta 1-4 glycosidic linka	iges		

		linkages					
	(C)	Highly branched, branched	(D)	Highly branched, branched sites			
		sites contain alpha 1-6		contain beta 1-6 glycosidic			
		glycosidic linkages		linkages			
MCQ 6	Which of the following pairs of sugars consists of anomers						
	(A)	d-glucose and d-fructose	(B)	α -d-glucose and β -d-glucose			
	(C)	d-galactose and d-glucose	(D)	d-glucose and d-mannose			
MCQ 7	The red precipitate formed when glucose is heated with "Benedict's reagent" is						
	(A)	Cupric hydroxide	(B)	Cuprous hydroxide			
	(C)	Cupric oxide	(D)	Cuprous oxide			
MCQ 8	One of the following does not have Sulfuric acid groups						
	(A)	Keratin sulphate	(B)	Chondroitin sulphate			
	(C)	Hyaluronic acid	(D)	Heparin			
MCQ 9	Example for "Fructosan" is						
	(A)	Starch	(B)	Inulin			
	(C)	Cellulose	(D)	Chitin			
MCQ 10	Chitin is polymer of						
	(A)	Glucose	(B)	N-Acetylglucosamine			
	(C)	Amylopectin	(D)	Hyaluronic acid			
MCQ 11	Example of trace element is						
	(A)	Κ	(B)	Mg			
	(C)	С	(D)	Ι			
MCQ 12	What type of biomolecule is an enzyme						
	(A)	Lipid	(B)	Protein			
	(C)	Carbohydrate	(D)	Nucleic acid			
MCQ 13	Which of the following tests distinguishes between aldoses and ketoses						
	(A)	Bial's Test	(B)	Barfoed's Test			
	· /	Seliwanoff's Test	(D)	Molisch's test			
MCQ 14	Which of the following is not a hexose						
	(A)	Fructose	(B)	Ribose			
	(C)	Mannose	(D)	Galactose			
MCQ 15	A sugar with 4 asymmetric carbon atoms will have following number of possible						
	isomers						
	(A)	8	(B)	32			
	(C)	16	(D)	4			

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