

PARUL UNIVERSITY
SCHOOL OF PHARMACY
B.PHARM FIRST SEMESTER

FIRST INTERNAL THEORY EXAMINATION: 2018-19

Subject Name: Pharmaceutical Inorganic Chemistry

Subject Code: BP104T

Time: 10:00 am to 11:30 am

Date: 03/10/2018

Total Marks: 30

Instructions:

1. Figures to the right indicate full marks.
2. Make suitable assumptions wherever necessary.

Q.1 Multiple Choice Questions:

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|------|---|-----------|
| (1) | In the limit test of Iron, Thioglycolic acid act as | 01 |
| | (a)Oxidizing agent (b)Suspending agent(c)Reducing agent (d)None of above | |
| (2) | _____Used as electrolyte replenisher to maintain electrolyte balance. | 01 |
| | (a) Nacl (b) H ₃ PO ₄ (c) Hcl (d) Ba ₃ (PO ₄) ₃ | |
| (3) | Calcium gluconate is prepared by | 01 |
| | (a)Lactic acid & CaCO ₃ (b)Oxalic acid & CaCO ₃ | |
| | (c)Gluconic acid & CaCO ₃ (d)Gluconic acid & Ca(OH) ₂ | |
| (4) | A ____%w/v solution of Nacl is isotonic with serum. | 01 |
| | (a) 0.8% w/v (b) 0.6% w/v (c) 0.9% w/v (d) 0.1% w/v | |
| (5) | First edition of pharmacopoeia was published in | 01 |
| | (a) 1965 (b) 1955 (c) 2000 (d) 1989 | |
| (6) | Iodine when reacts with starch gives | 01 |
| | (a)Red color (b)Blue color (c)Green color (d)Colorless | |
| (7) | Anaemia is | 01 |
| | (a)Deficiency of iron (b)Deficiency of zinc | |
| | (c)Deficiency of sodium chloride (d)None of above | |
| (8) | Apart from Expectorant, Ammonium Chloride is also used as | 01 |
| | (a)Haematinic (b)Acid-base equilibrium (c)Antimicrobial (d)None of the above | |
| (9) | Define :Haematinics | 01 |
| (10) | Define: Limit test | 01 |

Q.2 Long Answers: (Any One)

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|----|--|-----------|
| 1) | Write source of impurities in pharmaceutical substances. | 10 |
| 2) | Discuss the preparation, properties, assay, storage and uses of Ammonium chloride. | 10 |

Q.3 Short Answers: (Any Two)

- | | | |
|----|---|-----------|
| 1) | Define and classify expectorants. Explain mechanism of action of potassium iodide as expectorant. | 05 |
| 2) | Discuss the principle and reactions involved in limit test of sulphate. | 05 |
| 3) | Discuss Assay of any two: | 05 |
| | 1)Sodium Chloride 2)Copper Sulphate 3)Ferrous Sulphate | |