## **PARUL UNIVERSITY** FACULTY OF APPLIED SCIENCE M.Sc., Winter 2019-20 Examination

Enrollment No:\_\_\_\_\_

Semester: 1 Subject Code: 11203109 Subject Name: Biophysical Chemistry	Date: 27/11/2019 Time: 10:30 am to 01:00 pm Total Marks: 60	
Instructions:		
1. All questions are compulsory.		
2. Figures to the right indicate full marks.		
3. Make suitable assumptions wherever necessary.		
4. Start new question on new page.		
<b>O.1.</b> A) Essay type/ Brief note (4x2) (Each of 04 marks)		(08)
(a) Explain- Enzyme assays and controls.		
(b) Explain methods of Cell disintegration.		
Q.1. B) Answer the following questions (Any two)		
(a) Short note/ Brief note $(2x2)$ / Schematically label the figures (2	x2) (Each of 02 marks)	(04)
1. Application of Circular Dichroism		
2. Dialysis and Ultra-filtration		
(b) Short note – Plasma Emission Spectroscopy with diagrammati	c representation	(04)
(c) Short note – NMR and PMR with diagrammatic representation		(04)
Q.2. A) Answer the following questions.	-1)	(0.4)
(a) Short note/ Brief note (2x2)/ Fill in the blanks. (Each of 02 ma	irKS)	(04)
1. Explain - Chieffa for Protein purity and techniques for prot		
(b) Short note HPI C and EPI C with diagrammatic representation		(04)
(0) short note- in Le and Fi Le with diagrammatic representation O 2 B) Answer the following questions (Any two)	I	(04)
(a) Short note/ Multiple choice questions (Fach of 01 mark)		(03)
1. Gradient electrophoresis		(00)
2. Pulsed field gel electrophoresis		
3. Ion exchange chromatography		
(b) Short note – Theory and applications of polyacrylamide and ag	arose gel electrophoresis	(03)
(c) Short note – 2D Electrophoresis with diagrammatic representation	ion	(03)
Q.3. A) Essay type/ Brief note (4x2) (Each of 04 marks)		(08)
(a) Explain –Differential and Density gradient centrifugation		
(b) Explain- Preparative centrifugation and Ultra-centrifugation		
Q.3. B) Answer the following questions (Any two)	$\mathbf{O} = \mathbf{O} = $	
(a) Short note/ Brief note $(2x2)$ / Schematically label the figures (2	(Each of 02 marks)	(04)
1. KCF 2. Sodimentation Coefficient		
2. Sedimentation Coefficient		(04)
(c) Short note – Determination of molecular weight by sedimentar	tion velocity and equilibrium	(04)
methods	tion verberty and equilibrium	(04)
<b>O.4.</b> A) Answer the following questions.		
(a) Short note/ Brief note (2x2)/ Fill in the blanks. (Each of 02 ma	urks)	(04)
1. Principles of Spectroscopy	,	
2. Explain- Flow cytometry.		
(b) Short note – Blotting Techniques with diagrammatic representation	ation	(04)
Q.4. B) Answer the following questions (Any two)		
(a) Short note/ Multiple choice questions. (Each of 01 mark)		(03)
1. Beer-Lambert Law		
2. Flame Photometry		
3. Atomic absorption spectroscopy		
(b) Short note- western Blot with diagrammatic representation		(03)
(c) Snort note – I K Spectroscopy		(03)