PARUL UNIVERSITY FACULTY OF PHARMACY

Pharm.D November - 2019-20 Examination

Year: 1 Date: 29/11/2019

Subject Code: 08207105 Time: 10:00 am to 01:00pm

Subject Name: Pharmaceutical Inorganic Chemistry

Total Marks: 70

Instructions

- 1. Figures to the right indicate maximum marks.
- 2. Make suitable assumptions wherever necessary.

Q.1 Essay Type Questions. (any 2 out of 3) (15 Mark Each)

(30)

- 1. Define radioactivity. Mention the various units of radioactivity. Explain Geuger-Muller counter.
- 2. Explain the principle of acid-base titrations. Write a note on neutralization curves. Add a note on mixed indicators.
- 3. What is complexometric titration? Mention its application? Explain about masking and demasking agents.

Q.2 Short Essay Type Questions. (any 4 out of 5) (5 Mark Each)

(20)

- 1. List out various volumetric methods. Explain redox titration with suitable examples.
- 2. What is cerimetry? Mention about its advantages.
- 3. What are electrolyte replenishers? Write a note on ORS.
- 4. Explain the different theories of pH indicators.
- 5. Define accuracy and precision. How can the errors be minimized?

Q.3 Short Answers. (2 Mark Each)

(20)

- 1. Write the principle involved in Mohr's titration.
- 2. What is gravimetry? Enumerate the experimental techniques of gravimetric analysis.
- 3. Give a list of solvents and indicators used in non-aqueous titrations.
- 4. Why citric acid is added in the limit test of iron.
- 5. Explain the principle involved in the limit test for chlorides.
- 6. What are medicinal gases? How are they produced?
- 7. Define acidifiers with suitable examples.
- 8. Enlist ideal properties of antacids.
- 9. Define dentifrices? Explain the role of fluorides in dental caries.
- 10. Explain the physiological role of iodine.