Seat No:	Enrollment No:

# PARUL UNIVERSITY FACULTY OF PHARMACY Pharm.D Examination April - 2019

Year: 1 Date:24/04/2019

Subject Code: 08207105 Time:10:00 am to 01:00pm

Subject Name: Pharmaceutical Inorganic chemistry

Total Marks:70

# Instructions

- 1. Figures to the right indicate maximum marks.
- 2. Make suitable assumptions wherever necessary.

# Q.1 Essay Type Questions. (any 2 out of 3) (15 Mark Each)

(30)

- 1. Define buffer, derive Henderson-Hesselbach equation for pH and give a brief note on physiological acid base balance.
- 2. Enlist redox titrations. Explain theory of redox titration & Explain principle of any two redox titrations.
- 3. Define limit test. Write source of impurities and Explain principle of chloride and iron limit tests.

# Q.2 Short Essay Type Questions. (any 4 out of 5) (5 Mark Each)

(20)

- 1. Write in detail about non-aqueous solvents.
- 2. Explain in detail Mohrs method in precipitation titration.
- 3. Write in detail theory of complexometric titrations.
- 4. Enlist antacids & write ideal characteristics of Antacid.
- 5. Enlist types of detectors in radiopharmaceuticals. write about units of measurement of radioactivity.

#### **Q.3 Short Answers.** (2 Mark Each)

(20)

- 1. Write a brief note on electrolytes.
- 2. Write a brief note respiratory stimulants.
- 3. Classify catharatics. write in brief about magnesium sulphate.
- 4. Write a brief note on anasthetics.
- 5. Write a brief note on dentifrices.
- 6. Define following terms: Expectorants, Sclerosing agents
- 7. Write the uses of preservatives and filter aids.
- 8. Define following terms: Antimicrobials, antidotes.
- 9. Write chemical formula, synonyms, storage and uses of Sodium bicarbonate.
- 10. Define following terms: Emetics, sedatives