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PARUL UNIVERSITY

FACULTY OF PHARMACY

B. PHARM FIRST SEMESTER

SECOND INTERNAL THEORY EXAMINATION: 2022-23

SECOND INTERNAL TILLOW		
Subject 6	Name: Pharmaceutical Analysis Code: BP102 T 1.00pm to 01.15 min	Date: 21-02-2022 Total Marks: 30
1 \	take suitable assumptions wherever nece	essary.
, F	igures to the right indicate maximum m	arks.
		(10)
Q.1	Multiple Choice Questions.	
1	Mordant Black II indicator used in	b) Precipitation titration
	a) Complex metric titration	d) Acid-base titration
	c) Non-aqueous titration	d) Acid base than
2	End point colour in the Mohr method is	b) Blue colour
	a) Red colour c) Yellow colour	d) Black colour
	Adsorption Indicator are used in	
3	a) Fajan's method	b) Mohr's method
	c) Volhard's method	d) All
4	Reduction Involves	
•	a)) Loss of Oxygen	b) Gain in hydrogen
	c) Gain in oxygen	d) Loss of electron
5	Oxidizing agent does not include	
70	a) Potassium iodide	b) Bromine Solution
	c) Potassium dichromate	d) None of these
6	Oxidation Involve	
	a) Loss of hydrogen	b) Gain in hydrogen
	c) Gain in hydrogen	d) Gain in electron
7	Which of the following are redox titration	ons?
	a) Acid base Titration	b) Iodometry Titration
	c) Precipitation Titration	d) All of these
8	Hydrogen acts as reducing agent?	L) Du siving alastean
	a) By taking oxygen	b) By giving electron
	c) By taking Hydrogen	d) Both A and B
9		h) Dranning Margury
	a) Glass	b) Dropping Mercury
	c) Platinum	d) Silver
10		b) reference electrode
	a) Indicator electrode	d) None of the above
	c) Both A & B	d) None of the above

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PARUL UNIVERSITY

FACULTY OF PHARMACY

B. PHARM FIRST SEMESTER

SECOND INTERNAL THEORY EXAMINATION: 2022-23

Subject Name: Pharmaceutical Analysis

Subject Code: BP102 T Time: 12.00pm to 01.15 min

Date: 21-02-2022 Total Marks: 30

Instructions:

- 1. Make suitable assumptions wherever necessary.
- 2. Figures to the right indicate maximum marks.
- Q.2 Long Answer: (Answer Any one)

(10)

- 1) Enlist reference and indicator electrodes used in potentiometry titrations? Briefly Explain construction and working of reference electrode with diagram?
- 2) Write the principle of conductometric titration and explain type of conductometric titration with graph?
- Q.3 Short Answer: (Answer Any Two)

(10)

- 1) Write a short note on Volhard's method of precipitation titration?
- 2) What the principal complexometric titration? Difference between masking and demasking agent?
- 3) Difference between Iodometry and Iodimetry?