Roll	No.:	Enrolment No	
	PARUL U	<u>INIVERSITY</u>	
	FACULTY O	OF PHARMACY	
	B. PHARM FI	RST SEMESTER	
	IMPROVEMENT THEOI	RY EXAMINATION: 2022-23	
Sub			
Subject Name: Pharmaceutical Inorganic Che Subject Code: BP104T Time: 1 hr 15 min		Date: 01/03/2023 Total Marks: 30	
	ructions: . Make suitable assumptions wherever	necessary.	
	2. Figures to the right indicate maximum		
•	Multiple Choice Questions. $(10 \text{ X } 1 = 1)$		(10)
1	In limit test of sulphate, alcohol not allow		
	a) saturation	b) super saturation	
_	c) less saturation	d) precipitation	
2	Indian Pharmacopoeia is published by		
	a) International Pharmacopoeial	b) International Pharmacopoeial	
	committee	commission	
•	b) Indian Pharmacopoeial committee	d) Indian Pharmacopoeial commission	
3	Impurities in pharmaceutical preparation	•	
	a) Raw material	b) Manufacturing process	
4	c) Chemical instability The averall height of Nessler evilinder is	d) All of the above	
4	The overall height of Nessler cylinder is a) 110 mm		
	c) 190 mm	b) 150 mm d) 200 mm	
_	,	,	
3	In Gutzeit apparatus, which paper is used to trap any hydrogen sulphide which may have evolved along with arsine?		
	a) pH paper	b) Litmus paper	
	c) Lead acetate paper	d) Filter paper	
6	According to Lewis concept acid is	a, i itoi pupoi	
U	a) electron pair acceptor	b) electron pair donor	
	c) proton accepter	d) electron acceptor	
7	Selection of buffer depends upon	-, 	
-	a) Chemical factors	b) Pharmaceutical factors	

d) None of the above

b) Hyponatremia

d) Hypokalemia

b) Polishing agent

d) Desensitizing agent

c) The pH does not changes with even after addition of small quantities of acid or bases

c) Both a and b

d) All of the above

a) Hypernatremia

c) Hyperkalemia

a) Anticaries agent

c) Cleaning agent

10

Which is the properties of buffer

Sodium fluoride is an example of

a) The pH of buffer solution is constantb) The pH does not changes with dilution

Increased level of the sodium in the body is called as

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PARUL UNIVERSITY

FACULTY OF PHARMACY

B. PHARM FIRST SEMESTER

IMPROVEMENT THEORY EXAMINATION: 2022-23

Subject Name: Pharmaceutical Inorganic Chemistry

Subject Code: BP104T Date: 01/03/2023
Time: 1 hr 15 min Total Marks: 30

- Q.2 Long Answer: (Answer Any one) (1 X 10 = 10) (10)
- 1 Explain various acid-base theories.
- 2 Explain major intra and extra cellular electrolytes. Give a brief account on Oral Rehydration Salt.
- Q.3 Short Answer (Answer Any Two) (2 X 5=10) (10)
- 1 Define pharmacopoeia. Give a brief note on Indian Pharmacopoeia.
- 2 Enlist the sources of the impurities in pharmaceuticals and discuss the manufacturing hazards as sources of impurities.
- 3 Define limit test. Write a detailed note on the limit test for Iron.