PARUL UNIVERSITY FACULTY OF ENGINEERING & TECHNOLOGY Diploma Engineering, Mid semester Examination

Semester: 3 rd Date: (10/08/2022)Subject Code: (03613207)Time: (1hr: 30minSubject Name: (Basic Engineering Thermodynam)Total Marks: 40		(10/08/2022) (1hr: 30min)	
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Inst	tructions:		
1. A	Attempt all questions.		
2. N	Aake suitable assumptions wherever necessary.		
3. F	igures to the right indicate full marks.		
4. E	English version is considered to be Authentic.		
0.1	Answer any six out of Ten. (2 Marks Each)	(12)	
×	1. Define Entropy.	()	
	2. What is Specific Entropy? Give SI units.		
	3. Define Enthalpy with its related SI units.		
	4. Give Definition of Internal Energy.		
	5. What is Mechanical Equilibrium?		
	6. Give Definition of Thermodynamics.		
	7. Give Definition of Moveable Boundary with example.		
	8. Give definition of Absolute Zero Temperature with units.		
	9. Give definition of Specific Heat.		
	10. Give Statement of Conversion of Energy.		
Q.2	A) Explain Temperature Measurement Devices.	(03)	
	OR		
	A) Explain Shaft work with proper figure.	(03)	
	B) Give Different Statement of First Law of Thermodynamics.	(03)	
	OR		
	B) Explain Control Volume.	(03)	
	C) Which type thermodynamic system & boundary fluid flow in Pipe? Explain with figure.	(04)	
	C) Which type thermodynamic system & boundary boiled water in Filled Air in Ballon? Explain with	(04)	
	figure		
~ 1	D) Explain Joule's Experiments with proper Figure.	(04)	
Q.3	A) Explain Reversible Process with example.	(03)	
	OK	(0.2)	
	A) Explain Inteversible Process with example. B) What is Intensive Property? Cive Example	(03)	
	B) what is intensive Property? Give Example.	(03)	
	D) Droof Internal Energy is Droporty	(02)	
	D) Proof Internal Energy Is Property.	(03)	
	C) Derive Steady Flow Energy Equation.	(04)	
	C) Differentiate between Point Function and Path function	(04)	
	C) Differentiate Detween Fourt Function and Fain function.	(04) (04)	
	D) Classification of Thermodynamics System. Explain Isolated System with Example.	$(\mathbf{U} + \mathbf{J})$	